## Preparation and Electrochemical Properties of NiCo<sub>2</sub>O<sub>4</sub>/rGO Composites

Changjuan Yao, Yu Su<sup>\*</sup>, Ying Li, Jun Li

School of Materials Engineering, Shanghai University of Engineering Science, Shanghai, 201620, China. \*E-mail: suvu@sues.edu.cn

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NiCo<sub>2</sub>O<sub>4</sub>/rGO as electrode materials for supercapacitor were prepared by a facile hydrothermal route and co-precipitation. XRD and SEM were used to determine the lattice characteristics and morphology of NiCo<sub>2</sub>O<sub>4</sub>/rGO nanocomposite. It was found that GO had a good lamellar layer structure, and the overall structure was a flower-like structure. NiCo<sub>2</sub>O<sub>4</sub> in the two methods is uniformly attached to the surface of rGO. The shape of NiCo<sub>2</sub>O<sub>4</sub>/rGO nanocomposites prepared by solvothermal method (S-NiCo) after composite is similar to that of sea urchin, and the gap distribution between needles is reasonable. The NiCo<sub>2</sub>O<sub>4</sub>/rGO nanocomposites prepared by coprecipitation method (C-NiCo) have threedimensional network structure, and the nanocomposites are oriented. Using calomel electrode as reference electrode, a three electrode system was constructed for electrochemical experiments. The specific capacitance of NiCo<sub>2</sub>O<sub>4</sub>/rGO prepared by coprecipitation method is 1063.5F/g, that of NiCo<sub>2</sub>O<sub>4</sub>/rGO prepared by solvothermal method is 935.6F·g<sup>-1</sup>, and that of NiCo<sub>2</sub>O<sub>4</sub> prepared by coprecipitation method is 913.4F·g<sup>-1</sup> by constant current charge discharge test at 2A·g<sup>-1</sup>. The resistance measured by the AC impedance experiment is 1.13 $\Omega$ , which is less than 1.35 $\Omega$ of NiCo<sub>2</sub>O<sub>4</sub>/rGO electrode material prepared by solvothermal method.

Keywords: electrochemical, nanomaterials, supercapacitor, NiCo<sub>2</sub>O<sub>4</sub>/rGO material

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