

Anti-corrosive Properties of New Eco-friendly Dimethylamino Compounds on C- steel Corrosion in 2 M HCl

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The inhibitive effect of new eco-friendly dimethylamino compounds, named (2E,4E)-5-(4-(Dimethylamino)phenyl)-1-(pyridin-2-yl)penta-2,4-dien-1-one (Compound (I)) and (E)-1-(Anthracen-10-yl)-3-(4-(dimethylamino)phenyl)prop-2-en-1-one (Compound (II)) versus CS and its adsorption habit were obtained in 2 M HCl solution utilized Weight loss, TP, (EIS) and (EFM) techniques. The outcome obtained from IE improve with the raising the inhibitor dose. The %IE orders of organic compound are follow: Compound (II) > Compound (I). Kinetic activation parameters such as activation energy, entropy and enthalpy of activation were obtained from the influence of temperature on inhibition and corrosion. The types of new eco-friendly dimethylamino compounds are mixed kind inhibitor whose adsorption found by isotherm Temkin. EIS diagrams show that adsorption of new eco-friendly dimethylamino compounds increases the transfer resistance and decrease the capacitance of interface metal/solution. The morphology of inhibited CS was evaluated by SEM and EDX. Relation among calculations of quantum chemical and protection efficiency of the investigated assembled is confirmed by utilized the Density Functional Theory (DFT). Molecular docking was utilized to predict the binding among new eco-friendly dimethylamino compounds and the receptor of breast cancer mutant 3hb5-oxidoreductase.

Keywords: Eco-friendly, Adsorption, C-steel, SEM–EDX, Molecular docking.

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